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Objectives. By the end

of this section, you will

be able to: Determine

the shape of a

molecule using the

VSEPR. Draw the Lewis

structures of a

molecule including

bond angles and formal

charges.

Lab 11 Introduction |

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Chemistry I

Laboratory Manual

Laboratory 11:

Molecular Compounds
and Lewis Structures

Introduction. Molecular
compounds are formed
by sharing electrons

between non-metal
atoms. A useful theory
for understanding the
formation of molecular
compounds, shapes of
molecules and several
other properties is
called Lewis-dot

theory. We will explore

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the use of Lewis-dot theory to generate structures of molecular compounds and to build 3D models in order to predict shapes, and polarity of molecules.

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Determina PART 2:
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Molar Solubility and Ap
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made to be
broken—especially in
the chemistry lab. The
following rules exist for
your safety and should
always be followed.

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